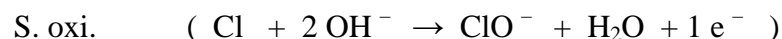
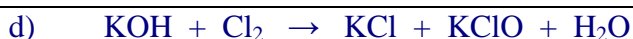
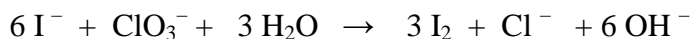
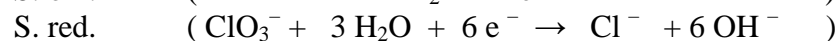
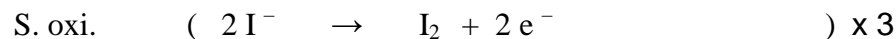
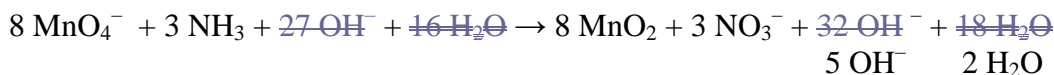
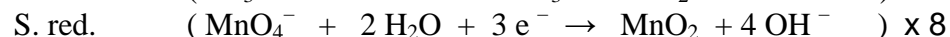
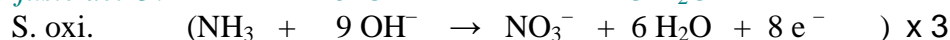
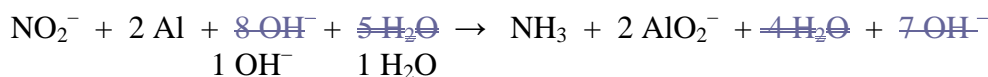
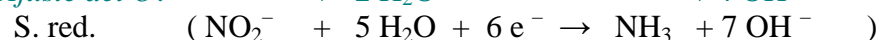
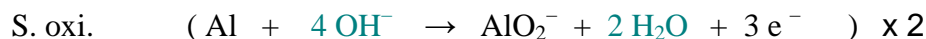
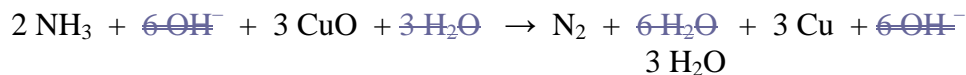
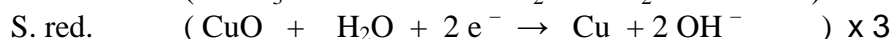
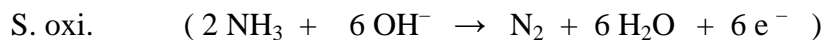
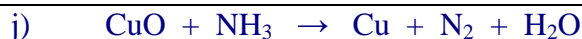
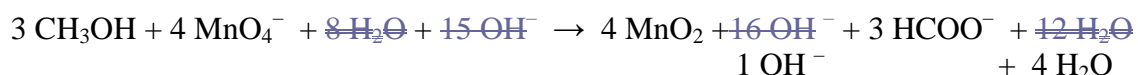
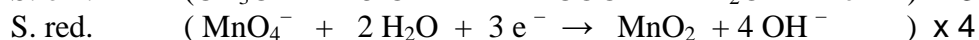
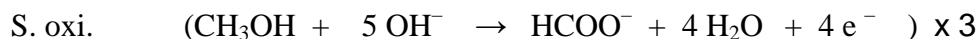
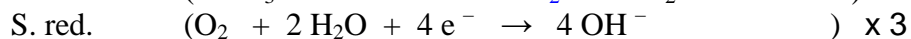
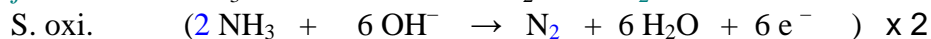
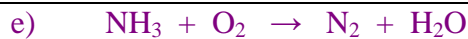


2.- Ajusta las siguientes reacciones por el método del ión-electrón en medio básico
Para ajustar los O, añadimos el doble de OH⁻ que oxígenos faltan en el lado con menos oxígeno, y el H₂O necesaria en el otro lado.

Si hay H, primero se ajusta éste, añadiendo tantos OH⁻ como H hay en exceso en el lado en que hay más hidrógeno, y el H₂O necesaria en el otro lado. A continuación se ajusta el O, como se indica en el primer punto.





(Se oxidan tanto el Cr^{3+} a Cr^{6+} , como el I^- a I^{7+} . Se reduce el Cl a Cl^-)

