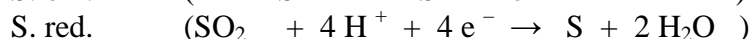
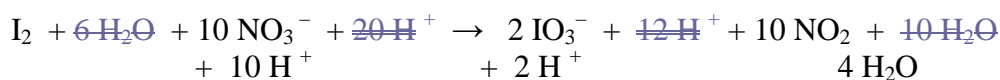
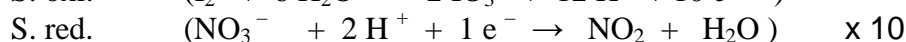
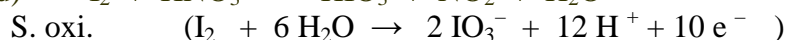
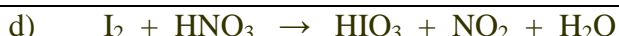
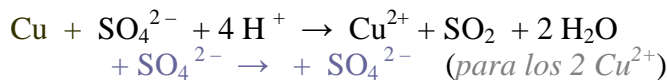
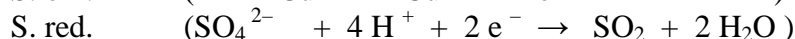
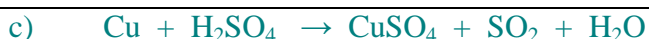
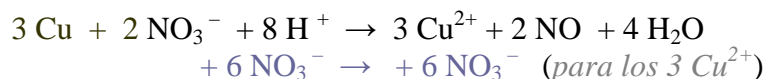
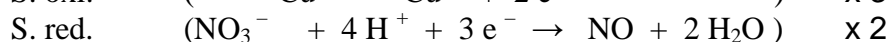
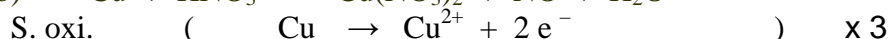
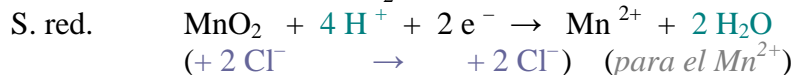
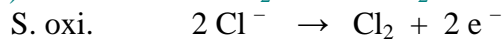
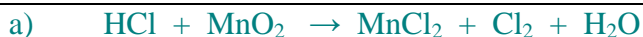
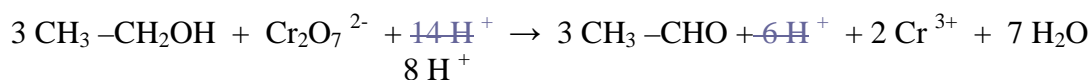
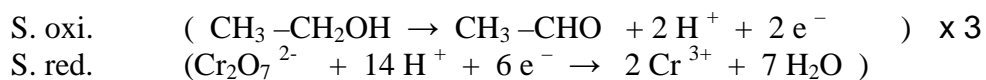
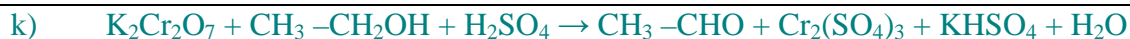
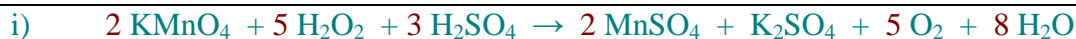
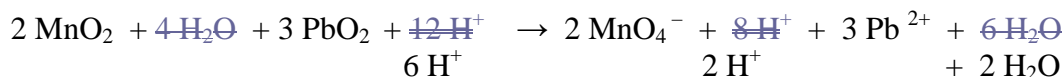
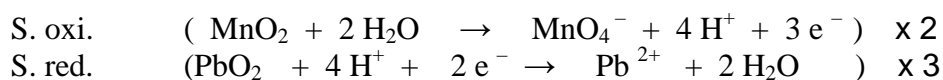
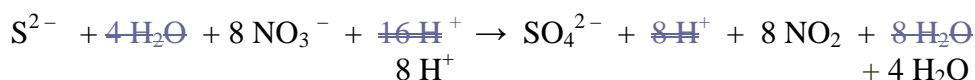
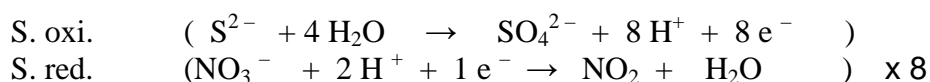
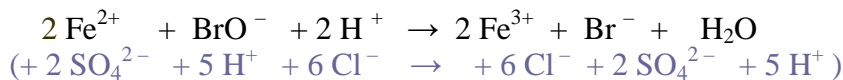
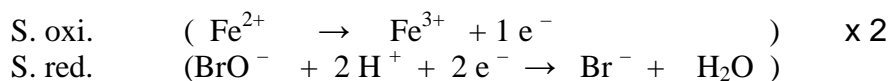
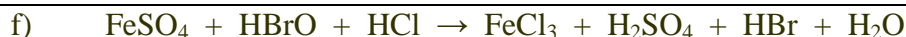


1.- Ajusta las siguientes reacciones por el método del ión-electrón en medio ácido  
*Para ajustar el oxígeno, se añade el H<sub>2</sub>O necesaria en el lado que falta O, y se ajusta con H<sup>+</sup> en el otro lado.*





*Ajustada:*



*(Se oxidan tanto el Cu<sup>+</sup> a Cu<sup>2+</sup>, como el S<sup>2-</sup> a S<sup>4+</sup>. Se reduce el N<sup>5+</sup> a N<sup>2+</sup>)*

